Scroll down for all Safety Data Sheets (SDS) for this product.

Total Enclosures: 3



Simplicity in Water Analysis

Cover Page for Safety Data Sheet

Thank you for choosing CHEMetrics, Inc. We appreciate your business. In order to best serve your needs for accurate and complete Safety Data, we offer the following information as supplemental to the attached SDS.

SDS No.: R1605

Version No.: 2.5

Product Name: Bromine, Chlorine, Chlorine Dioxide, Hydrogen Peroxide, Ozone and Peracetic Acid CHEMets® Refills & Vacu-vials® Ampoules, Peracetic Acid VACUettes Refills, Hypochlorite CHEMets® Refill

Part Nos.: R-1605, K-1613 Ampoules, R-2500, R-2504, R-2509, K-2513 Ampoules, K-2523 Ampoules, K-2703 Ampoules, R-2705, R-5502, K-5513 Ampoules, R-5808, R-7404, K-7423 Ampoules, R-7904, R-7904A, R-7904B, R-7904C, R-7904D, K-7913 Ampoules

Product Descriptions:

CHEMets Refills: Sealed glass ampoules, 7 mm OD, for visual colorimetric water analysis. Each CHEMet[™] ampoule contains approximately 0.22 mL of liquid reagent sealed under vacuum. Refills contain 30 ampoules, test kits contain 1 refill.

VACUettes Refills: Sealed glass ampoules, 7 mm OD, with small glass capillary attached, for visual colorimetric water analysis. Each VACUette[™] ampoule contains approximately 0.22 mL of liquid reagent sealed under vacuum. Refills contain 30 ampoules, test kits contain 1 refill.

Vacu-vials Ampoules: Sealed glass ampoules, 13 mm OD, for instrumental colorimetric water analysis. Each Vacu-vial[™] ampoule contains approximately 0.8 mL of liquid reagent sealed under vacuum. Test kits contain 30 ampoules.

Addendum to Section 14 Transport Information:

Shipping container markings and labels for this product, as received, may vary from the contents of section 14 of the SDS for one or both of the following reasons:

- CHEMetrics has packaged this product as Dangerous Goods in Excepted Quantities according to IATA, US DOT, and IMDG regulations.
- CHEMetrics has packaged this product as part of a test kit or reagent set composed of various chemical reagents and elected to ship as UN 3316 Chemical Kit, Hazard Class 9, Packing Group II or III.

In case of reshipment, it is the responsibility of the shipper to determine appropriate labels and markings in accordance with applicable transportation regulations.

Additional Information:

- "Print Date" = Revision Date (expressed as DD/MM/YYY)
- Test kits and reagents sets may contain additional chemical reagents. See separate SDS(s).

CHEMets®, VACUettes®, Vacu-vials®, and Titrets® are registered trademarks of CHEMetrics Inc.



CHEMetrics, Inc.

Chemwatch: 9-87748 SDS No: R1605 Version No: 2.5 Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

Chemwatch Hazard Alert Code: 0

Issue Date: 19/11/2014 Print (Revision) Date: 09/04/2018 S.GHS.USA.EN

SECTION 1 IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

Product Identifier

Product name	Bromine, Chlorine, Chlorine Dioxide, Hydrogen Peroxide, Ozone, Peracetic Acid CHEMets Refills & Vacu-vials Ampoules, Peracetic Acid VACUettes Refills, Hypochlorite CHEMets Refill
Synonyms	Part Nos.: R-1605, K-1613 Ampoules, R-2500, R-2504, R-2509, K-2513 Ampoules, K-2523 Ampoules, K-2703 Ampoules, R-2705, R-5502, K-5513 Ampoules, R-5808, R-7404, K-7423 Ampoules, R-7904, R-7904A, R-7904B, R-7904C, R-7904D, K-7913 Ampoules
Proper shipping name	Not Applicable
Chemical formula	Not Applicable
Other means of identification	Not Available
CAS number	Not Applicable

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	Component of water analysis test kits I-2001, I-2005, I-2019, I-2020, K-1605, K-1613, K-2500, K-2504, K-2504A, K-2504B, K-2504C, K-2504D, K-2513,
	K-2523, K-2703, K-2705, K-5502, K-5513, K-5808, K-5816, K-7404, K-7423, K-7904,K-7904A, K-7904B, K-7904C, K-7904D, K-7913

Details of the manufacturer/importer

Registered company name	CHEMetrics, Inc.
Address	4295 Catlett Road, Midland, VA. 22728 United States
Telephone	1-540-788-9026
Fax	1-540-788-4856
Website	www.chemetrics.com
Email	technical@chemetrics.com

Emergency telephone number

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Association / Organisation	ChemTel Inc.
Emergency telephone numbers	1-800-255-3924
Other emergency telephone numbers	+01-813-248-0585

SECTION 2 HAZARDS IDENTIFICATION

Classification of the substance or mixture

GHS Classification	Flammable Liquid Category 4
Label elements	
GHS label elements	Not Applicable
SIGNAL WORD	WARNING
Hazard statement(s)	
H227	Combustible liquid
Precautionary statement(s)) Prevention
P101	If medical advice is needed, have product container or label at hand.
P102	Keep out of reach of children.
P103	Read label before use.
P210	Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking.

P280	Wear protective gloves/protective clothing/eye protection/face protection.	
Precautionary statement(s)) Response	
P370+P378	In case of fire: Use to extinguish.	
Precautionary statement(s)	Storage	
P403+P235	Store in a well-ventilated place. Keep cool.	
Precautionary statement(s	Precautionary statement(s) Disposal	
P501	Dispose of contents/container to authorised chemical landfill or if organic to high temperature incineration	

SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
7732-18-5	>94	water
67-64-1	<4	acetone
7778-77-0	1	potassium phosphate, monobasic
62637-92-7	0.1	N,N-diethyl-p-phenylenediamine oxalate
Not Available	<1	Proprietary ingredient
Not Available	<0.1	Proprietary ingredient

SECTION 4 FIRST AID MEASURES

Description of first aid measures

Eye Contact	If this product comes in contact with eyes: Wash out immediately with water. If irritation continues, seek medical attention. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	If skin or hair contact occurs: Flush skin and hair with running water (and soap if available). Seek medical attention in event of irritation.
Inhalation	 If fumes, aerosols or combustion products are inhaled remove from contaminated area. Other measures are usually unnecessary.
Ingestion	 Immediately give a glass of water. First aid is not generally required. If in doubt, contact a Poisons Information Centre or a doctor.

Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

SECTION 5 FIREFIGHTING MEASURES

Extinguishing media

	 ▶ Water spray or fog. ▶ Foam.
	 Foam. Dry chemical powder. BCF (where regulations permit).
	Carbon dioxide.
pecial hazards arising fro	om the substrate or mixture

Fire Incompatibility None known.

Advice for firefighters

Fire Fighting	 Alert Fire Brigade and tell them location and nature of hazard. Wear full body protective clothing with breathing apparatus. Prevent, by any means available, spillage from entering drains or water course. Use water delivered as a fine spray to control fire and cool adjacent area. Avoid spraying water onto liquid pools.
Fire/Explosion Hazard	 Combustible. Slight fire hazard when exposed to heat or flame. Heating may cause expansion or decomposition leading to violent rupture of containers. On combustion, may emit irritating/ toxic fumes. May emit acrid smoke.

SECTION 6 ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

Minor Spills	 Remove all ignition sources. Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes. Control personal contact with the substance, by using protective equipment. Contain and absorb spill with sand, earth, inert material or vermiculite.
Major Spills	 Moderate hazard. Clear area of personnel and move upwind. Alert Fire Brigade and tell them location and nature of hazard. Wear breathing apparatus plus protective gloves. Prevent, by any means available, spillage from entering drains or water course.
	Personal Protective Equipment advice is contained in Section 8 of the MSDS.

SECTION 7 HANDLING AND STORAGE

Precautions for safe handling

Safe handling	 Avoid all personal contact, including inhalation. Wear protective clothing when risk of exposure occurs. Use in a well-ventilated area. Prevent concentration in hollows and sumps. DO NOT enter confined spaces until atmosphere has been checked. Wear impact- and splash-resistant eyewear. Break the ampoule tip only when it is completely immersed in sample. Breaking the tip in air may cause the glass ampoule to shatter.
Other information	 Store in original containers. Keep containers securely sealed. No smoking, naked lights or ignition sources. Store in a cool, dry, well-ventilated area. Store away from incompatible materials and foodstuff containers. For optimum analytical performance, store in the dark and at room temperature.

Conditions for safe storage, including any incompatibilities

Suitable container	Metal can or drum Packaging as recommended by manufacturer. Check all containers are clearly labelled and free from leaks.
Storage incompatibility	None known

PACKAGE MATERIAL INCOMPATIBILITIES

Not Available

SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

Control parameters

OCCUPATIONAL EXPOSURE LIMITS (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
US OSHA Permissible Exposure Levels (PELs) - Table Z1	acetone	Acetone	2400 mg/m3 / 1000 ppm	Not Available	Not Available	Not Available
US ACGIH Threshold Limit Values (TLV)	acetone	‡ Acetone	500 ppm	750 ppm	Not Available	TLV® Basis: (URT & eye irr CNS impair; hematologic eff); BEI
US NIOSH Recommended Exposure Limits (RELs)	acetone	Dimethyl ketone, Ketone propane, 2-Propanone	590 mg/m3 / 250 ppm	Not Available	Not Available	Not Available

EMERGENCY LIMITS

Ingredient	Material name TEEL-			TEEL-2	TEEL-3	
acetone	Acetone	Not Availa	Ible	Not Available	Not Available	
potassium phosphate, monobasic	Potassium phosphate, monobasic	sphate, monobasic 29 mg/m3		320 mg/m3	1900 mg/m3	
Ingredient	Original IDLH		Revised IDLH			
water	Not Available		Not Available			
acetone	20,000 ppm		2,500 [LEL] ppm	2,500 [LEL] ppm		
potassium phosphate, monobasic	Not Available		Not Available	Not Available		
N,N-diethyl- p-phenylenediamine oxalate	Not Available		Not Available			
Proprietary ingredient	Not Available		Not Available			
Proprietary ingredient	Not Available		Not Available			

Exposure controls	
Appropriate engineering controls	Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk. Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed property.
Personal protection	
Eye and face protection	 Safety glasses with side shields Chemical goggles. Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available.
Skin protection	See Hand protection below
Hands/feet protection	The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application. The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice. Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include:
Body protection	See Other protection below
Other protection	 Overalls. P.V.C. apron. Barrier cream. Skin cleansing cream.
Thermal hazards	Not Available

Recommended material(s)

GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the: "Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the computergenerated selection:

Bromine, Chlorine, Chlorine Dioxide, Hydrogen Peroxide, Ozone, Peracetic Acid, Hypochlorite Refills and Ampoules

Material	СРІ
BUTYL	А
BUTYL/NEOPRENE	С
CPE	С
HYPALON	С
NAT+NEOPR+NITRILE	С
NATURAL RUBBER	С
NATURAL+NEOPRENE	С
NEOPRENE	С
NEOPRENE/NATURAL	С
NITRILE	С
NITRILE+PVC	С
PE	С
PE/EVAL/PE	С
PVA	С
PVC	С
PVDC/PE/PVDC	С
SARANEX-23	С
SARANEX-23 2-PLY	С
TEFLON	С
VITON	С
VITON/CHLOROBUTYL	С
VITON/NEOPRENE	С

Respiratory protection

Type AX Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Selection of the Class and Type of respirator will depend upon the level of breathing zone contaminant and the chemical nature of the contaminant. Protection Factors (defined as the ratio of contaminant outside and inside the mask) may also be important.

Required minimum protection factor	Maximum gas/vapour concentration present in air p.p.m. (by volume)	Half-face Respirator	Full-Face Respirator
up to 10	1000	AX-AUS / Class1	-
up to 50	1000	-	AX-AUS / Class 1
up to 50	5000	Airline *	-
up to 100	5000	-	AX-2
up to 100	10000	-	AX-3
100+			Airline**

* - Continuous Flow ** - Continuous-flow or positive pressure demand

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

* CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

NOTE: As a series of factors will influence the actual performance of the glove, a final

selection must be based on detailed observation. -

* Where the glove is to be used on a short term, casual or infrequent basis, factors such as

"feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise

be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Appearance	Colorless to pale pink		
Appearance			
Physical state	Liquid	Relative density (Water = 1)	1.0
Odour	Odourless	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	6.3	Decomposition temperature	Not Available
Melting point / freezing point (°C)	-7	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	94	Molecular weight (g/mol)	Not Available
Flash point (°C)	>75	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Combustible.	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water (g/L)	Miscible	pH as a solution	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

SECTION 10 STABILITY AND REACTIVITY

Reactivity	See section 7
Chemical stability	 Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 TOXICOLOGICAL INFORMATION

Information on toxicological effects

Inhaled	The material is not thought to produce adverse health effects or irritation of the respiratory tract (as classified by EC Directives using animal models). Nevertheless, good hygiene practice requires that exposure be kept to a minimum and that suitable control measures be used in an occupational setting. Not normally a hazard due to non-volatile nature of product
Ingestion	The material has NOT been classified by EC Directives or other classification systems as "harmful by ingestion". This is because of the lack of corroborating animal or human evidence.
Skin Contact	The material is not thought to produce adverse health effects or skin irritation following contact (as classified by EC Directives using animal models). Nevertheless, good hygiene practice requires that exposure be kept to a minimum and that suitable gloves be used in an occupational setting. Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.
Eye	Although the liquid is not thought to be an irritant (as classified by EC Directives), direct contact with the eye may produce transient discomfort characterised by tearing or conjunctival redness (as with windburn).
Chronic	Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure.

IRRITATION

TOXICITY			

CHEMets Refills, VACUettes Refills, Vacu-vials Ampoules	ΤΟΧΙΟΙΤΥ	IRRITATION	

ACETONE	The material may cause skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin. for acetone: The acute toxicity of acetone is low. Acetone is not a skin irritant or sensitiser but is a defatting agent to the skin. Acetone is an eye irritant. The subchronic toxicity of acetone has been examined in mice and rats that were administered acetone in the drinking water and again in rats treated by oral gavage.
POTASSIUM PHOSPHATE, MONOBASIC	No data of toxicological significance identified in literature search.
N,N-DIETHYL- P-PHENYLENEDIAMINE OXALATE	The following information refers to contact allergens as a group and may not be specific to this product. Contact allergies quickly manifest themselves as contact eczema, more rarely as urticaria or Quincke's oedema. The pathogenesis of contact eczema involves a cell-mediated (T lymphocytes) immune reaction of the delayed type. Other allergic skin reactions, e.g. contact urticaria, involve antibody- mediated immune reactions. The significance of the contact allergen is not simply determined by its sensitisation potential: the distribution of the substance and the opportunities for contact with it are equally important.
WATER	No significant acute toxicological data identified in literature search.

Acute Toxicity	0	Carcinogenicity	0
Skin Irritation/Corrosion	\otimes	Reproductivity	0
Serious Eye Damage/Irritation	0	STOT - Single Exposure	0
Respiratory or Skin sensitisation	0	STOT - Repeated Exposure	0
Mutagenicity	0	Aspiration Hazard	0

Legend:

Data required to make classification available
 Data available but does not fill the criteria for classification

S – Data Not Available to make classification

CMR STATUS

Not Applicable

SECTION 12 ECOLOGICAL INFORMATION

Toxicity

DO NOT discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
water	LOW	LOW
acetone	LOW (Half-life = 14 days)	MEDIUM (Half-life = 116.25 days)
Proprietary ingredient	LOW	LOW

Bioaccumulative potential

Ingredient	Bioaccumulation
water	LOW (LogKOW = -1.38)
acetone	LOW (BCF = 69)
Proprietary ingredient	LOW (LogKOW = -3.8796)

Mobility in soil

Ingredient	Mobility
water	LOW (KOC = 14.3)
acetone	HIGH (KOC = 1.981)
Proprietary ingredient	LOW (KOC = 14.3)

SECTION 13 DISPOSAL CONSIDERATIONS

Waste treatment methods

Product / Packaging disposal

Dispose of according to federal, state, and local regulations.

SECTION 14 TRANSPORT INFORMATION

Labels Required	
Marine Pollutant	NO
Land transport (DOT): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS	
Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS	

Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

SECTION 15 REGULATORY INFORMATION

Safety, health and environmental regulations / legislation specific for the substance or mixture

water(7732-18-5) is found on the following regulatory lists	"US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory"	
acetone(67-64-1) is found on the following regulatory lists	Recommended Exposure Limits (RELs)", "US - Alaska Limits for Air Contaminants", "US - Washington Permissible exposure limits of air contaminants", "US	
potassium phosphate, monobasic(7778-77-0) is found on the following regulatory lists	"US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory"	
N,N-diethyl- p-phenylenediamine oxalate(62637-92-7) is found on the following regulatory lists	"US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory"	
Proprietary ingredient() is found on the following regulatory lists		
Proprietary ingredient() is found on the following regulatory lists	"Not Applicable"	

SECTION 16 OTHER INFORMATION

Other information

Ingredients with multiple cas numbers

Name	CAS No
Not Available	Not Available

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

A list of reference resources used to assist the committee may be found at:

www.chemwatch.net/references

The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

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Simplicity in Water Analysis

Cover Page for Safety Data Sheet

Thank you for choosing CHEMetrics, Inc. We appreciate your business. In order to best serve your needs for accurate and complete Safety Data, we offer the following information as supplemental to the attached SDS.

SDS No.: S1600

Version No.: 1.2

Product Name: Activator Solution for Bromine, Chlorine, Hydrogen Peroxide, Ozone, Peracetic Acid Test Kits

Part Nos.: A-1600, A-2500, A-2500V, A-5501, A-7400, A-7900

Product Descriptions:

Activator Solution: Plastic bottle, contains approximately 9 mL of liquid reagent. Test kits contain one (1) bottle of solution. Activator Solution packs contain six (6) bottles of solution.

Addendum to Section 14 Transport Information:

Shipping container markings and labels for this product, as received, may vary from the contents of section 14 of the SDS for one or both of the following reasons:

- CHEMetrics has packaged this product as Dangerous Goods in Excepted Quantities according to IATA, US DOT, and IMDG regulations.
- CHEMetrics has packaged this product as part of a test kit or reagent set composed of various chemical reagents and elected to ship as UN 3316 Chemical Kit, Hazard Class 9, Packing Group II or III.

In case of reshipment, it is the responsibility of the shipper to determine appropriate labels and markings in accordance with applicable transportation regulations.

Additional Information:

- "Print Date" = Revision Date (expressed as DD/MM/YYYY)
- Test kits and reagents sets may contain additional chemical reagents. See separate SDS(s).

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Activator Solution for Bromine, Chlorine, Hydrogen Peroxide, Ozone, Peracetic Acid Test Kits

CHEMetrics, Inc.

Chemwatch: **9-76789** SDS No: **S1600** Version No: **1.2** Safety Data Sheet according to OSHA HazCom Standard (2012) requirements Chemwatch Hazard Alert Code: 2 Issue Date: 05/09/2014 Print Date: 12/03/2015 Initial Date: 06/09/2014 S.GHS.USA.EN

SECTION 1 IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

Product Identifier

Product name Activator Solution for Bromine, Chlorine, Hydrogen Peroxide, Ozone, Peracetic Acid Test Kits	
Synonyms	Part Nos.: A-1600, A-2500, A-2500V, A-5501, A-7400, A-7900
Proper shipping name	Not Applicable
Chemical formula	Not Applicable
Other means of identification	Not Available
CAS number	Not Applicable

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	Component of water analysis test kits I-2001, I-2019, I-2020, K-1605, K-1613, K-2500, K-2504, K-2504A, K-2504B, K-2504C, K-2504D, K-2505, K-2505V,
	K-2511, K-2513, K-5502, K-5513, K-7402, K-7404, K-7423, K-7904, K-7913

Details of the manufacturer/importer

Registered company name	CHEMetrics, Inc.
Address	4295 Catlett Road, Midland, VA. 22728 United States
Telephone	1-540-788-9026
Fax	1-540-788-4856
Website	www.chemetrics.com
Email	technical@chemetrics.com

Emergency telephone number

Association / Organisation	ChemTel Inc.
Emergency telephone numbers	1-800-255-3924
Other emergency telephone numbers	+01-813-248-0585

SECTION 2 HAZARDS IDENTIFICATION

Classification of the substance or mixture

GHS Classification	Skin Corrosion/Irritation Category 2, Eye Irritation Category 2A, Respiratory Sensitizer Category 1, Skin Sensitizer Category 1, Reproductive Toxicity Category 2, STOT - SE (Resp. Irr.) Category 3
Label elements	
GHS label elements	
SIGNAL WORD	DANGER
Hazard statement(s)	
H315	Causes skin irritation
H319	Causes serious eye irritation
H334	May cause allergy or asthma symptoms or breathing difficulties if inhaled
H317	May cause an allergic skin reaction
H361	Suspected of damaging fertility or the unborn child

Activator Solution for Bromine, Chlorine, Hydrogen Peroxide, Ozone, Peracetic Acid Test Kits

H335 May cause respiratory irritation

Precautionary statement(s) Prevention			
P101	P101 If medical advice is needed, have product container or label at hand.		
P102	2 Keep out of reach of children.		
P103	Read label before use.		
P201	Obtain special instructions before use.		
P261	Avoid breathing dust/fume/gas/mist/vapours/spray.		
P271	Use only outdoors or in a well-ventilated area.		
P280	Wear protective gloves/protective clothing/eye protection/face protection.		
P284	[In case of inadequate ventilation] wear respiratory protection.		

Precautionary statement(s) Response

P304+P340	INHALED: Remove person to fresh air and keep comfortable for breathing.				
100411040					
P308+P313	IF exposed or concerned: Get medical advice/attention.				
P342+P311	experiencing respiratory symptoms: Call a POISON CENTER/doctor/physician/first aider				
P302+P352	IF ON SKIN: Wash with plenty of water and soap				
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.				

Precautionary statement(s) Storage

P405	Store locked up.	
P403+P233	Store in a well-ventilated place. Keep container tightly closed.	

Precautionary statement(s) Disposal

P501 Dispos

Dispose of contents/container to authorised chemical landfill or if organic to high temperature incineration

SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name		
7681-11-0	11	potassium iodide		
1310-73-2	<0.1	sodium hydroxide		
7732-18-5	>88	water		

SECTION 4 FIRST AID MEASURES

Description of first aid measures

Eye Contact	 If this product comes in contact with the eyes: Wash out immediately with fresh running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Seek medical attention without delay; if pain persists or recurs seek medical attention. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	 If skin contact occurs: Immediately remove all contaminated clothing, including footwear. Flush skin and hair with running water (and soap if available). Seek medical attention in event of irritation.
Inhalation	 If fumes or combustion products are inhaled remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. Transport to hospital, or doctor, without delay.
Ingestion	 Immediately give a glass of water. First aid is not generally required. If in doubt, contact a Poisons Information Centre or a doctor.

Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

SECTION 5 FIREFIGHTING MEASURES

Extinguishing media

There is no restriction on the type of extinguisher which may be used.
Use extinguishing media suitable for surrounding area.

Special hazards arising from the substrate or mixture

Fire Incompatibility	None known.
Advice for firefighters	

Fire Fighting	 Alert Fire Brigade and tell them location and nature of hazard. Wear breathing apparatus plus protective gloves in the event of a fire. Prevent, by any means available, spillage from entering drains or water courses. Use fire fighting procedures suitable for surrounding area. DO NOT approach containers suspected to be hot.
Fire/Explosion Hazard	 Non combustible. Not considered a significant fire risk, however containers may burn. Decomposition may produce toxic fumes of; hydrogen iodideMay emit poisonous fumesMay emit corrosive fumes.

SECTION 6 ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

Minor Spills	 Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes. Control personal contact with the substance, by using protective equipment. Contain and absorb spill with sand, earth, inert material or vermiculite. Wipe up.
Major Spills	Moderate hazard. Clear area of personnel and move upwind. Alert Fire Brigade and tell them location and nature of hazard. Wear breathing apparatus plus protective gloves. Prevent, by any means available, spillage from entering drains or water course.
	Personal Protective Equipment advice is contained in Section 8 of the MSDS.

SECTION 7 HANDLING AND STORAGE

Precautions for safe handling

Safe handling	 Avoid all personal contact, including inhalation. Wear protective clothing when risk of exposure occurs. Use in a well-ventilated area. Prevent concentration in hollows and sumps. DO NOT enter confined spaces until atmosphere has been checked. Wear impact- and splash-resistant eyewear. 	
Other information	information For optimum analytical performance, store in the dark and at room temperature.	

Conditions for safe storage, including any incompatibilities

Suitable container	 Polyethylene or polypropylene container. Packing as recommended by manufacturer. Check all containers are clearly labelled and free from leaks.
Storage incompatibility	None known

PACKAGE MATERIAL INCOMPATIBILITIES

Not Available

SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

Control parameters

OCCUPATIONAL EXPOSURE LIMITS (OEL)

INGREDIENT DATA						
Source Ingredient		Material name	TWA	STEL	Peak	Notes
US ACGIH Threshold Limit Values (TLV)	potassium iodide	lodine and lodides - lodides	0.01 ppm	0.1 ppm	Not Available	TLV® Basis: Hypothyroidism; URT irr
US OSHA Permissible Exposure Levels (PELs) - Table Z1	sodium hydroxide	Sodium hydroxide	2 mg/m3	Not Available	Not Available	Not Available
US ACGIH Threshold Limit Values (TLV)	sodium hydroxide	Sodium hydroxide	Not Available	Not Available	2 mg/m3	TLV® Basis: URT, eye, & skin irr
US NIOSH Recommended Exposure Limits (RELs)	sodium hydroxide	Caustic soda, Lye, Soda Iye, Sodium hydrate	Not Available	Not Available	2 mg/m3	Not Available

EMERGENCY LIMITS

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
potassium iodide	Potassium iodide	0.18 mg/m3	1.9 mg/m3	140 mg/m3
sodium hydroxide	Sodium hydroxide	Not Available	Not Available	Not Available

Activator Solution for Bromine, Chlorine, Hydrogen Peroxide, Ozone, Peracetic Acid Test Kits

Ingredient	Original IDLH	Revised IDLH
potassium iodide	Not Available	Not Available
sodium hydroxide	250 mg/m3	10 mg/m3
water	Not Available	Not Available

Exposure controls

Appropriate engineering controls	Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk. Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed property.
Personal protection	
Eye and face protection	 Safety glasses with side shields. Chemical goggles. Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience.
Skin protection	See Hand protection below
Hands/feet protection	 Wear chemical protective gloves, e.g. PVC. Wear safety footwear or safety gumboots, e.g. Rubber NOTE: The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact. Contaminated leather items, such as shoes, belts and watch-bands should be removed and destroyed. The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer.
Body protection	See Other protection below
Other protection	 Overalls. P.V.C. apron. Barrier cream. Skin cleansing cream.
Thermal hazards	Not Available

Recommended material(s)

GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the $\ computer-$

generated selection: Activator Solution for Bromine, Chlorine, Hydrogen Peroxide, Ozone, Peracetic Acid Test Kits

Material	CPI
BUTYL	С
NAT+NEOPR+NITRILE	С
NATURAL RUBBER	С
NATURAL+NEOPRENE	С
NEOPRENE	С
NEOPRENE/NATURAL	С
NITRILE	С
NITRILE+PVC	С
PE	С
PE/EVAL/PE	С
PVA	С
PVC	С
SARANEX-23 2-PLY	С
SARANEX-23	С
TEFLON	С
VITON	С
VITON/CHLOROBUTYL	С

* CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

NOTE: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

* Where the glove is to be used on a short term, casual or infrequent basis, factors such as

Respiratory protection

Particulate. (AS/NZS 1716 & 1715, EN 143:000 & 149:001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required.

Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	-AUS P2	-	-PAPR-AUS / Class 1 P2
up to 50 x ES	-	-AUS / Class 1 P2	-
up to 100 x ES	-	-2 P2	-PAPR-2 P2 ^

^ - Full-face

 $\begin{array}{l} \mbox{A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC) \\ \end{array}$

Activator Solution for Bromine, Chlorine, Hydrogen Peroxide, Ozone, Peracetic Acid Test Kits

"feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Physical state Liquid Relative density (Water = 1) >1 Odour Odourless Partition coefficient n-octanol / water Not Available Odour threshold Not Available Auto-ignition temperaturer (°C) Not Available pH (as supplied) 11 Decomposition temperature Not Available od viscosity (cst) Not Available filtial boiling point / freezing point (°C) c0 Viscosity (cst) Not Available Initial boiling range (°C) 100 Molecular weight (g/moi) Not Available Flash point (°C) Not Applicable Explosive properties Not Available Upper Explosive Limit (%) Not Available Surface Tension (dy/mcm m/Mm) Not Available Lower Explosive Limit (%) Not Available Surface Tension (dy/mcm m/Mm) Not Available Vapour pressure (kPa) Not Available Gas group Not Available Vapour density (Air = 1) Not Available Gas group Not Available	Appearance	colorless to pale yellow		
OdourOdourlessPartition coefficient n-octanol / waterNot AvailableOdour thresholdNot AvailableAuto-ignition temperature (°C)Not AvailablepH (as supplied)11Decomposition temperatureNot AvailableMelting point / freezing point (°C)<0Viscosity (cst)Not AvailableInitial boiling noint and boiling range (°C)100Molecular weight (g/mol)Not AvailableFlash point (°C)Not ApplicableTasteNot AvailableEvaporation rate Upper Explosive Limit (%)Not AvailableOxidising propertiesNot AvailableLower Explosive Limit (%)Not AvailableSurface Tension (d/m/cm or m/N/m)Not AvailableVapour pressure (kPa)Not AvailableGas groupNot AvailableSolubility in water (g/L)MiscibleMetaleFlam a solutionNot Available				
OdourOdourlessNot AvailableNot AvailableOdour thresholdNot AvailableAuto-ignition temperature (°C)Not AvailablePH (as supplied)11Decomposition temperatureNot AvailableMelting point / freezing point (°C)c0Viscosity (SS)Not AvailableInitial boiling point and boiling range (°C)100Molecular weight (g/mol)Not AvailableFlash point (°C)Not ApplicableExplosive (g/mol)Not AvailableViscosity (SS)Not AvailableNot AvailableUpper Explosive Limit (%)Not AvailableSurface Tension (dyn/cm or mN/m)Vapour pressure (kPa)Not AvailableVolatile Component (%vool)Not AvailableVapour pressure (kPa)Not AvailableGas groupNot AvailableSolubility in water (g/L)MisciblePH as a solutionNot Available	Physical state	Liquid	Relative density (Water = 1)	>1
Odour thresholdNot AvailableNot AvailablepH (as supplied)11Decomposition temperatureNot AvailableMelting point / freezing point (*C)<0Viscosity (cSt)Not AvailableInitial boiling point and boiling range (*C)100Molecular weight (g/mol)Not AvailableFlash point (*C)Not ApplicableTasteNot AvailableEvaporation rateNot AvailableExplosive propertiesNot AvailableInitial boiling range (*C)Not AvailableOxidising propertiesNot AvailableUpper Explosive Limit (%)Not AvailableSurface Tension (dyn/cm or mN/m)Not AvailableVapour pressure (kPa)Not AvailableVolatile Component (%vol)Not AvailableVapour pressure (kPa)Not AvailableGas groupNot AvailableSolubility in water (g/L)MiscibleMisciblePH as a solutionNot Available	Odour	Odourless		Not Available
PH (as supplied)11ItemperatureNot AvailableMelting point / freezing point (°C)<0Viscosity (cSt)Not AvailableInitial boiling point and boiling range (°C)100Molecular weight (g/mol)Not AvailableFlash point (°C)Not ApplicableTasteNot AvailableEvaporation rateNot AvailableExplosive propertiesNot AvailableFlammabilityNot AvailableOxidising propertiesNot AvailableUpper Explosive Limit (%)Not AvailableSurface Tension (dyn/cm or mN/m)Not AvailableLower Explosive Limit (%)Not AvailableVolatile Component (%vol)Not AvailableSolubility in water (g/L)MiscibleMisciblePH as a solutionNot Available	Odour threshold	Not Available		Not Available
Point (°C)<0	pH (as supplied)	11	•	Not Available
boiling range (°C)100Molecular weight (g/mol)Not AvailableFlash point (°C)Not ApplicableTasteNot AvailableEvaporation rateNot AvailableExplosive propertiesNot AvailableFlammabilityNot ApplicableOxidising propertiesNot AvailableUpper Explosive Limit (%)Not AvailableSurface Tension (dyn/cm or mN/m)Not AvailableLower Explosive Limit (%)Not AvailableVolatile Component (%vol)Not AvailableVapour pressure (kPa)Not AvailableOf AvailableNot AvailableSolubility in water (g/L)Not AvailableOf AvailableNot Available	•••••••	<0	Viscosity (cSt)	Not Available
Evaporation rateNot AvailableExplosive propertiesNot AvailableFlammabilityNot ApplicableOxidising propertiesNot AvailableUpper Explosive Limit (%)Not AvailableSurface Tension (dyn/cm or mN/m)Not AvailableLower Explosive Limit (%)Not AvailableVolatile Component (%vol)Not AvailableVapour pressure (kPa)Not AvailableGas groupNot AvailableSolubility in water (g/L)MisciblePH as a solutionNot Available		100	Molecular weight (g/mol)	Not Available
FlammabilityNot ApplicableOxidising propertiesNot AvailableUpper Explosive Limit (%)Not AvailableSurface Tension (dyn/cm or mN/m)Not AvailableLower Explosive Limit (%)Not AvailableVolatile Component (%vol)Not AvailableVapour pressure (kPa)Not AvailableGas groupNot AvailableSolubility in water (g/L)MisciblePH as a solutionNot Available	Flash point (°C)	Not Applicable	Taste	Not Available
Upper Explosive Limit (%)Not AvailableSurface Tension (dyn/cm or mN/m)Not AvailableLower Explosive Limit (%)Not AvailableVolatile Component (%vol)Not AvailableVapour pressure (kPa)Not AvailableGas groupNot AvailableSolubility in water (g/L)MisciblepH as a solutionNot Available	Evaporation rate	Not Available	Explosive properties	Not Available
Opper Explosive Limit (%) Not Available Not Available Not Available Lower Explosive Limit (%) Not Available Volatile Component (%vol) Not Available Vapour pressure (kPa) Not Available Gas group Not Available Solubility in water (g/L) Miscible PH as a solution Not Available	Flammability	Not Applicable	Oxidising properties	Not Available
Vapour pressure (kPa) Not Available Gas group Not Available Solubility in water (g/L) Miscible pH as a solution Not Available	Upper Explosive Limit (%)	Not Available		Not Available
Solubility in water (g/L) Miscible pH as a solution Not Available	Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
	Vapour pressure (kPa)	Not Available	Gas group	Not Available
Vapour density (Air = 1) Not Available VOC g/L Not Available	Solubility in water (g/L)	Miscible	pH as a solution	Not Available
	Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

SECTION 10 STABILITY AND REACTIVITY

Reactivity	See section 7
Chemical stability	 Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 TOXICOLOGICAL INFORMATION

Information on toxicological effects

Inhaled	The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage. Not normally a hazard due to non-volatile nature of product The material has NOT been classified by EC Directives or other classification systems as "harmful by inhalation". This is because of the lack of corroborating animal or human evidence.
Ingestion	The material has NOT been classified by EC Directives or other classification systems as "harmful by ingestion". This is because of the lack of corroborating animal or human evidence.
Skin Contact	This material can cause inflammation of the skin on contact in some persons. The material may accentuate any pre-existing dermatitis condition Skin contact is not thought to have harmful health effects (as classified under EC Directives); the material may still produce health damage following entry through wounds, lesions or abrasions. Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.
Eye	This material can cause eye irritation and damage in some persons.
Chronic	Long-term exposure to respiratory irritants may result in disease of the airways involving difficult breathing and related systemic problems. Inhaling this product is more likely to cause a sensitisation reaction in some persons compared to the general population. Skin contact with the material is more likely to cause a sensitisation reaction in some persons compared to the general population. Based on experience with animal studies, exposure to the material may result in toxic effects to the development of the foetus, at levels which do not cause significant toxic effects to the mother. Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure.
Activator Solution for Bromine,Chlorine, Hydrogen Peroxide, Ozone, Peracetic Acid Test Kits	TOXICITY IRRITATION

Activator Solution for			
Bromine,Chlorine, Hydrogen Peroxide, Ozone,	TOXICITY IRRITATI	ИС	
Peracetic Acid Test Kits			
	The material may produce severe irritation to the eye causing pronounced inflammation. R conjunctivitis. The material may cause severe skin irritation after prolonged or repeated exposure and m vesicles, scaling and thickening of the skin. Repeated exposures may produce severe ulce	ay produce on contact skin redness, swelling, the production of	
	Asthma-like symptoms may continue for months or even years after exposure to the materi	al ceases.	
WATER	No significant acute toxicological data identified in literature search.		
Activator Solution for Bromine,Chlorine, Hydrogen Peroxide, Ozone, Peracetic Acid Test Kits, POTASSIUM IODIDE	The following information refers to contact allergens as a group and may not be specific to Contact allergies quickly manifest themselves as contact eczema, more rarely as urticaria involves a cell-mediated (T lymphocytes) immune reaction of the delayed type. Other aller mediated immune reactions. The significance of the contact allergen is not simply determine and the opportunities for contact with it are equally important.	or Quincke's oedema. The pathogenesis of contact eczema gic skin reactions, e.g. contact urticaria, involve antibody-	
Acute Toxicity	© Carcinogeni	sity 🛇	
Skin Irritation/Corrosion	✓ Reproductiv		
Serious Eye Damage/Irritation	✓ STOT - Single Expos	ıre 🗸	
Respiratory or Skin sensitisation	✓ STOT - Repeated Exposition	ure 🛇	
Mutagenicity	S Aspiration Haze	ard 🛇	
CMR STATUS	Legend:	 Data required to make classification available Data available but does not fill the criteria for classification Data Not Available to make classification 	
EYE	sodium hydroxide US - California OEHHA/ARB - Acute Reference Exposure Levels a	nd Target Organs (RELs) - Eye X SODIUM HYDROXIDE	
RESPIRATORY	sodium hydroxide US - California OEHHA/ARB - Acute Reference Exposure Levels a	nd Target Organs (RELs) - Respiratory	

SECTION 12 ECOLOGICAL INFORMATION

SKIN

sodium hvdroxide

Toxicity

For lodine: lodine is an important element in studies of environmental protection and human health, global-scale hydrologic processes and nuclear non-proliferation. Inorganic and organic species that may be hydrophilic, atmophilic, and biophilic.

US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs) - Skin

Terrestrial Fate: There was an appreciable iodate reduction to iodide, presumably mediated by the structural iron(II), in some clay minerals. Humic acid in soil promotes the electrochemical reduction of iodine(12) to ionic iodide. The different oxidation species of iodine have markedly different sorption properties, hence, changes in iodine redox states can greatly affect the mobility of iodine in the environment.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
potassium iodide	HIGH	HIGH
sodium hydroxide	LOW	LOW
water	LOW	LOW

Bioaccumulative potential

Ingredient	Bioaccumulation
potassium iodide	LOW (LogKOW = 0.0436)
sodium hydroxide	LOW (LogKOW = -3.8796)
water	LOW (LogKOW = -1.38)

Mobility in soil

Ingredient	Mobility
potassium iodide	LOW (KOC = 14.3)
sodium hydroxide	LOW (KOC = 14.3)
water	LOW (KOC = 14.3)

SECTION 13 DISPOSAL CONSIDERATIONS

Х

Activator Solution for Bromine, Chlorine, Hydrogen Peroxide, Ozone, Peracetic Acid Test Kits

Waste treatment methods		
Product / Packaging disposal	Product / Packaging disposal Dispose of according to federal, state, and local regulations.	
SECTION 14 TRANSPORT INFORMATION		
Labels Required		
Marine Pollutant	NO	

Land transport (DOT): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

SECTION 15 REGULATORY INFORMATION

Safety, health and environmental regulations / legislation specific for the substance or mixture

potassium iodide(7681-11-0) is found on the following regulatory lists	"US ACGIH Threshold Limit Values (TLV) - Carcinogens", "US ACGIH Threshold Limit Values (TLV)", "US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory"
sodium hydroxide(1310-73-2) is found on the following regulatory lists	"US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants", "US - Idaho - Limits for Air Contaminants", "US - Hawaii Air Contaminant Limits", "US - California Permissible Exposure Limits for Chemical Contaminants", "US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants", "US - Oregon Permissible Exposure Limits (Z-1)", "US - Michigan Exposure Limits for Air Contaminants", "US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values", "US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs)", "US NIOSH Recommended Exposure Limits (RELs)", "US - Alaska Limits for Air Contaminants", "US - Washington Permissible exposure Limits of air contaminants", "US - Minnesota Permissible Exposure Limits (PELs)", "US ACGIH Threshold Limit Values (TLV)", "US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants", "US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory", "US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants", "US OSHA Permissible Exposure Levels (PELs) - Table Z1"
water(7732-18-5) is found on the following regulatory lists	"US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory"

SECTION 16 OTHER INFORMATION

Other information

Ingredients with multiple cas numbers

Name	CAS No
Not Available	Not Available

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

A list of reference resources used to assist the committee may be found at:

www.chemwatch.net/references

The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

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Simplicity in Water Analysis

Cover Page for Safety Data Sheet

Thank you for choosing CHEMetrics, Inc. We appreciate your business. In order to best serve your needs for accurate and complete Safety Data, we offer the following information as supplemental to the attached SDS.

SDS No.: S5500

Version No.: 2.2

Product Name: Activator Solution for Hydrogen Peroxide CHEMets® and Vacu-vials® Kits

Part Nos.: A-5500

Product Descriptions:

Activator Solution: Plastic bottle, contains approximately 9 mL of liquid reagent. Test kits contain one (1) bottle of solution. Activator Solution packs contain six (6) bottles of solution.

Addendum to Section 14 Transport Information:

Shipping container markings and labels for this product, as received, may vary from the contents of section 14 of the SDS for one or both of the following reasons:

- CHEMetrics has packaged this product as Dangerous Goods in Excepted Quantities according to IATA, US DOT, and IMDG regulations.
- CHEMetrics has packaged this product as part of a test kit or reagent set composed of various chemical reagents and elected to ship as UN 3316 Chemical Kit, Hazard Class 9, Packing Group II or III.

In case of reshipment, it is the responsibility of the shipper to determine appropriate labels and markings in accordance with applicable transportation regulations.

Additional Information:

- "Print Date" = Revision Date (expressed as DD/MM/YYY)
- Test kits and reagents sets may contain additional chemical reagents. See separate SDS(s).

CHEMets®, VACUettes®, Vacu-vials®, and Titrets® are registered trademarks of CHEMetrics Inc.



CHEMetrics, Inc.

Chemwatch: 9-90601 SDS No: 55500 Version No: 2.2 Safety Data Sheet according to OSHA HazCom Standard (2012) requirements Chemwatch Hazard Alert Code: 4

Issue Date: 04/09/2014 Print Date: 12/03/2015 Initial Date: 06/09/2014 S.GHS.USA.EN

SECTION 1 IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

Product Identifier

Product name	Activator Solution for Hydrogen Peroxide CHEMets and Vacu-vials Kits
Synonyms	Part No.: A-5500
Proper shipping name	Corrosive liquids, n.o.s. (contains sulfuric acid)
Chemical formula	Not Applicable
Other means of identification	Not Available
CAS number	Not Applicable

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses Component of water analysis test kits K-5502, K-5504, K-5513

Details of the manufacturer/importer

Registered company name	CHEMetrics, Inc.
Address	4295 Catlett Road, Midland, VA. 22728 United States
Telephone	1-540-788-9026
Fax	1-540-788-4856
Website	www.chemetrics.com
Email	technical@chemetrics.com

Emergency telephone number

Association / Organisation	ChemTel Inc.
Emergency telephone numbers	1-800-255-3924
Other emergency telephone numbers	+01-813-248-0585

SECTION 2 HAZARDS IDENTIFICATION

Classification of the substance or mixture

GHS Classification	Metal Corrosion Category 1, Skin Corrosion/Irritation Category 1A, Serious Eye Damage Category 1, Skin Sensitizer Category 1, STOT - SE (Resp. Irr.) Category 3, STOT - RE Category 2	
_abel elements		
GHS label elements		
SIGNAL WORD	DANGER	
lazard statement(s)		
H290	May be corrosive to metals	
H314	Causes severe skin burns and eye damage	
H318	Causes serious eye damage	
H317	May cause an allergic skin reaction	
H335	May cause respiratory irritation	

H373 May cause damage to organs through prolonged or repeated exposure

Precautionary statement(s) Prevention	
P101	If medical advice is needed, have product container or label at hand.
P102	Keep out of reach of children.
P103	Read label before use.
P260	Do not breathe dust/fume/gas/mist/vapours/spray.
P271	Use only outdoors or in a well-ventilated area.
P280	Wear protective gloves/protective clothing/eye protection/face protection.
P234	Keep only in original container.
P272	Contaminated work clothing should not be allowed out of the workplace.

Precautionary statement(s) Response

P301+P330+P331	30+P331 IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.	
P303+P361+P353	P303+P361+P353 IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water/shower.	
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.	
P310	P310 Immediately call a POISON CENTER/doctor/physician/first aider	
P302+P352	IF ON SKIN: Wash with plenty of water and soap	

Precautionary statement(s) Storage

P405	Store locked up.	
P403+P233	Store in a well-ventilated place. Keep container tightly closed.	

Precautionary statement(s) Disposal

P501 Dispose

Dispose of contents/container to authorised chemical landfill or if organic to high temperature incineration

SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
7732-18-5	88	water
7664-93-9	10	sulfuric acid
12027-67-7	2	ammonium heptamolybdate

SECTION 4 FIRST AID MEASURES

Description of first aid measures

Eye Contact	If this product comes in contact with the eyes: Immediately hold eyelids apart and flush the eye continuously with running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes. Transport to hospital or doctor without delay. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.	
Skin Contact	 If skin or hair contact occurs: Immediately flush body and clothes with large amounts of water, using safety shower if available. Quickly remove all contaminated clothing, including footwear. Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre. Transport to hospital, or doctor. 	
Inhalation	 If fumes or combustion products are inhaled remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. Transport to hospital, or doctor, without delay. Inhalation of vapours or aerosols (mists, fumes) may cause lung oedema. Corrosive substances may cause lung damage (e.g. lung oedema, fluid in the lungs). As this reaction may be delayed up to 24 hours after exposure, affected individuals need complete rest (preferably in semi-recumbent posture) and must be kept under medical observation, the administration of a spray containing a dexamethasone derivative or beclomethasone derivative may be considered. This must definitely be left to a doctor or person authorised by him/her. (ICSC13719) 	
Ingestion	 For advice, contact a Poisons Information Centre or a doctor at once. Urgent hospital treatment is likely to be needed. If swallowed do NOT induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. Observe the patient carefully. 	

- Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.
- Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.
- Transport to hospital or doctor without delay.

Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

- For acute or short term repeated exposures to strong acids:
- Airway problems may arise from laryngeal edema and inhalation exposure. Treat with 100% oxygen initially.
- Respiratory distress may require cricothyroidotomy if endotracheal intubation is contraindicated by excessive swelling
- Intravenous lines should be established immediately in all cases where there is evidence of circulatory compromise.
- Strong acids produce a coagulation necrosis characterised by formation of a coagulum (eschar) as a result of the dessicating action of the acid on proteins in specific tissues.

INGESTION:

Immediate dilution (milk or water) within 30 minutes post ingestion is recommended.

- DO NOT attempt to neutralise the acid since exothermic reaction may extend the corrosive injury.
- Be careful to avoid further vomit since re-exposure of the mucosa to the acid is harmful. Limit fluids to one or two glasses in an adult.
- · Charcoal has no place in acid management.
- Some authors suggest the use of lavage within 1 hour of ingestion.

SKIN:

• Skin lesions require copious saline irrigation. Treat chemical burns as thermal burns with non-adherent gauze and wrapping.

 Deep second-degree burns may benefit from topical silver sulfadiazine. EYE:

Eye injuries require retraction of the eyelids to ensure thorough irrigation of the conjuctival cul-de-sacs. Irrigation should last at least 20-30 minutes. DO NOT use neutralising agents or any other additives. Several litres of saline are required.

- Cycloplegic drops, (1% cyclopentolate for short-term use or 5% homatropine for longer term use) antibiotic drops, vasoconstrictive agents or artificial tears may be indicated dependent on the severity of the injury.
- Steroid eye drops should only be administered with the approval of a consulting ophthalmologist).

[Ellenhorn and Barceloux: Medical Toxicology]

SECTION 5 FIREFIGHTING MEASURES

Extinguishing media

 Dry chemical powder. BCF (where regulations permit). Carbon dioxide. 	Water spray or fog. Foam.
 BCF (where regulations permit). Carbon dioxide. 	Dry chemical powder.
► Carbon dioxide.	 BCF (where regulations permit).
	▶ Carbon dioxide.

Special hazards arising from the substrate or mixture

Fire Incompatibility None known.

Advice for firefighters

Fire Fighting	 Alert Fire Brigade and tell them location and nature of hazard. Wear full body protective clothing with breathing apparatus. Prevent, by any means available, spillage from entering drains or water course. Use fire fighting procedures suitable for surrounding area. Do not approach containers suspected to be hot.
Fire/Explosion Hazard	 Non combustible. Not considered to be a significant fire risk. Acids may react with metals to produce hydrogen, a highly flammable and explosive gas. Heating may cause expansion or decomposition leading to violent rupture of containers. May emit corrosive, poisonous fumes.

SECTION 6 ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

Minor Spills	 Drains for storage or use areas should have retention basins for pH adjustments and dilution of spills before discharge or disposal of material. Check regularly for spills and leaks. Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes. Control personal contact with the substance, by using protective equipment.
Major Spills	 Clear area of personnel and move upwind. Alert Fire Brigade and tell them location and nature of hazard. Wear full body protective clothing with breathing apparatus. Prevent, by any means available, spillage from entering drains or water course. Consider evacuation (or protect in place).
	Personal Protective Equipment advice is contained in Section 8 of the MSDS.

SECTION 7 HANDLING AND STORAGE

Precautions for safe handling

Safe handling	 Avoid all personal contact, including inhalation. Wear protective clothing when risk of exposure occurs. Use in a well-ventilated area. WARNING: To avoid violent reaction, ALWAYS add material to water and NEVER water to material. Avoid smoking, naked lights or ignition sources.
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	Wear impact- and splash-resistant eyewear.
Other information	 Store in original containers. Keep containers securely sealed. Store in a cool, dry, well-ventilated area. Store away from incompatible materials and foodstuff containers. Protect containers against physical damage and check regularly for leaks. For optimum analytical performance, store in the dark and at room temperature.

Conditions for safe storage, including any incompatibilities

Suitable container	 DO NOT use aluminium or galvanised containers Check regularly for spills and leaks Lined metal can, lined metal pail/ can. Plastic pail. Polyliner drum. Packing as recommended by manufacturer. Check all containers are clearly labelled and free from leaks.
Storage incompatibility	 Inorganic acids are generally soluble in water with the release of hydrogen ions. The resulting solutions have pH's of less than 7.0. Inorganic acids neutralise chemical bases (for example: amines and inorganic hydroxides) to form salts - neutralisation can generate dangerously large amounts of heat in small spaces. The dissolution of inorganic acids in water or the dilution of their concentrated solutions with additional water may generate significant heat. The addition of water to inorganic acids often generates sufficient heat in the small region of mixing to cause some of the water to boil explosively.

PACKAGE MATERIAL INCOMPATIBILITIES

Not Available

SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

Control parameters

OCCUPATIONAL EXPOSURE LIMITS (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
US OSHA Permissible Exposure Levels (PELs) - Table Z1	sulfuric acid	Sulfuric acid	1 mg/m3	Not Available	Not Available	Not Available
US ACGIH Threshold Limit Values (TLV)	sulfuric acid	Sulfuric acid	0.2 mg/m3	Not Available	Not Available	TLV® Basis: Pulm func
US NIOSH Recommended Exposure Limits (RELs)	sulfuric acid	Battery acid, Hydrogen sulfate, Oil of vitriol, Sulfuric acid (aqueous)	1 mg/m3	Not Available	Not Available	Not Available
US OSHA Permissible Exposure Levels (PELs) - Table Z1	ammonium heptamolybdate	Molybdenum - Soluble compounds	5 mg/m3	Not Available	Not Available	(as Mo)
US NIOSH Recommended Exposure Limits (RELs)	ammonium heptamolybdate	Synonyms vary depending upon the specific soluble molybdenum compound.	Not Available	Not Available	Not Available	See Appendix D

EMERGENCY LIMITS

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3	
sulfuric acid	Sulfuric acid	Not Available	Not Available Not Available Not Available		
ammonium heptamolybdate	Molybdic acid, hexaammonium salt; (Ammonium heptamolybdate)	2.6 mg/m3	29 mg/m3	170 mg/m3	
Ingredient	Original IDLH	Revised IDLH	Revised IDLH		
water	Not Available	Not Available	Not Available		
sulfuric acid	80 mg/m3	15 mg/m3			
ammonium heptamolybdate	N.E. mg/m3 / N.E. ppm	1,000 mg/m3			

Exposure controls

Appropriate engineering controls	Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk. Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly.
Personal protection	
Eye and face protection	 Safety glasses with unperforated side shields may be used where continuous eye protection is desirable, as in laboratories; spectacles are not sufficient where complete eye protection is needed such as when handling bulk-quantities, where there is a danger of splashing, or if the material may be under pressure. Chemical goggles.whenever there is a danger of the material coming in contact with the eyes; goggles must be properly fitted. Full face shield (20 cm, 8 in minimum) may be required for supplementary but never for primary protection of eyes; these afford face protection. Alternatively a gas mask may replace splash goggles and face shields. Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants.

Skin protection	See Hand protection below
Hands/feet protection	 Elbow length PVC gloves When handling corrosive liquids, wear trousers or overalls outside of boots, to avoid spills entering boots. NOTE: The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact. Contaminated leather items, such as shoes, belts and watch-bands should be removed and destroyed.
Body protection	See Other protection below
Other protection	 Overalls. PVC Apron. PVC protective suit may be required if exposure severe. Eyewash unit. Ensure there is ready access to a safety shower.
Thermal hazards	Not Available

Recommended material(s)

GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index". The effect(s) of the following substance(s) are taken into account in the *computer*-

generated selection: Activator Solution for Hydrogen Peroxide CHEMets and Vacu-vials Kits

Material CPI NEOPRENE A BUTYL С NATURAL RUBBER С С NATURAL+NEOPRENE С NEOPRENE/NATURAL NITRILE С PE С С PVA С PVC SARANEX-23 С VITON С

Respiratory protection

Type E-P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required.

Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	E-AUS P2	-	E-PAPR-AUS / Class 1 P2
up to 50 x ES	-	E-AUS / Class 1 P2	-
up to 100 x ES	-	E-2 P2	E-PAPR-2 P2 ^

^ - Full-face

 $\begin{array}{l} \mbox{A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC) \\ \end{array}$

* CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

NOTE: As a series of factors will influence the actual performance of the glove, a final

selection must be based on detailed observation. -

* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Appearance	Colorless to pale yellow		
Physical state	Liquid	Relative density (Water = 1)	1.06
Odour	Odourless	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	<1	Decomposition temperature	Not Available
Melting point / freezing point (°C)	0	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	120	Molecular weight (g/mol)	Not Available
Flash point (°C)	Not Applicable	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Applicable	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water (g/L)	Miscible	pH as a solution	Not Available

Vapour density (Air = 1)	Not Available	VOC g/L	Not Available
SECTION 10 STABILITY A	AND REACTIVITY		
Reactivity	See section 7		
Chemical stability	 Contact with alkaline material liberates heat Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur. 		
Possibility of hazardous reactions	See section 7		
Conditions to avoid	See section 7		
Incompatible materials	See section 7		
Hazardous decomposition products	See section 5		

SECTION 11 TOXICOLOGICAL INFORMATION

Information on toxicological effects

Inhaled	The material can cause respiratory irritation in some persons Corrosive acids can cause irritation of the respiratory tract, wit nausea and weakness. Not normally a hazard due to non-volatile nature of product The material has NOT been classified by EC Directives or ot	th coughing, choking and mucous m	embrane damage. There may be dizziness, headache,
Ingestion	Ingestion of acidic corrosives may produce burns around and speaking may also be evident. The material has NOT been classified by EC Directives or ot animal or human evidence. Molybdenum, an essential trace element, can in large doses I	her classification systems as "harmf	ul by ingestion". This is because of the lack of corroborating
Skin Contact	Skin contact with acidic corrosives may result in pain and burn Skin contact is not thought to have harmful health effects (as through wounds, lesions or abrasions. Open cuts, abraded or irritated skin should not be exposed to Entry into the blood-stream, through, for example, cuts, abrasis of the material and ensure that any external damage is suitably	classified under EC Directives); the i this material ions or lesions, may produce system	• • •
Eye	If applied to the eyes, this material causes severe eye damage Direct eye contact with acid corrosives may produce pain, tea completely.		d burns of the epithelia generally recover rapidly and
Chronic	Repeated or prolonged exposure to acids may result in the er and inflammation of lung tissue often occurs. Substance accumulation, in the human body, is likely and may Long-term exposure to respiratory irritants may result in disea Skin contact with the material is more likely to cause a sensitis	cause some concern following repe ase of the airways involving difficult b	ated or long-term occupational exposure. reathing and related systemic problems.
Activator Solution for Hydrogen Peroxide CHEMets and Vacu-vials Kits	ΤΟΧΙCΙΤΥ	IRRITATION	
Activator Solution for Hydrogen Peroxide CHEMets and Vacu-vials Kits	ΤΟΧΙCΙΤΥ	IRRITATION	
Activator Solution for Hydrogen Peroxide CHEMets and Vacu-vials Kits	The following information refers to contact allergens as a g Contact allergies quickly manifest themselves as contact er involves a cell-mediated (T lymphocytes) immune reaction of mediated immune reactions. The significance of the contact and the opportunities for contact with it are equally important	czema, more rarely as urticaria or Qu of the delayed type. Other allergic sk t allergen is not simply determined b	uincke's oedema. The pathogenesis of contact eczema
WATER	No significant acute toxicological data identified in literature	e search.	
SULFURIC ACID	Occupational exposures to strong inorganic acid mists of s	sulfuric acid:	
SULFURIC ACID, AMMONIUM HEPTAMOLYBDATE	Asthma-like symptoms may continue for months or even years as reactive airways dysfunction syndrome (RADS) which c diagnosis of RADS include the absence of preceding respir within minutes to hours of a documented exposure to the irr bronchial hyperreactivity on methacholine challenge testing in the criteria for diagnosis of RADS. RADS (or asthma) for of and duration of exposure to the irritating substance.	an occur following exposure to high ratory disease, in a non-atopic indivic itant. A reversible airflow pattern, on and the lack of minimal lymphocytic	levels of highly irritating compound. Key criteria for the dual, with abrupt onset of persistent asthma-like symptoms spirometry, with the presence of moderate to severe inflammation, without eosinophilia, have also been included
Acute Toxicity	0	Carcinogenicity	0

S - Data Not Available to make classification

Activator Solution for Hydrogen Peroxide CHEMets and Vacu-vials Kits

Serious Eye Damage/Irritation	v	STOT - Single Exposure	*
Respiratory or Skin sensitisation	*	STOT - Repeated Exposure	*
Mutagenicity	0	Aspiration Hazard	0
			 Data required to make classification available Data available but does not fill the criteria for classification

CMR STATUS

CARCINOGEN	sulfuric acid US Environmental Defense Scorecard Recognized Carcinogens P65	
RESPIRATORY	sulfuric US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs) - Respiratory US - California acid OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELs) - Respiratory	x

SECTION 12 ECOLOGICAL INFORMATION

Toxicity

Ecotoxicity:

The tolerance of water organisms towards pH margin and variation is diverse. Recommended pH values for test species listed in OECD guidelines are between 6.0 and almost 9. Acute testing with fish showed 96h-LC50 at about pH 3.5

For Molybdenum:

Environmental Fate: Molybdenum is an essential micronutrient in plants and animals. It is commonly used in the manufacture of steel alloys. Based on the high concentration of molybdenum in all analyzed waste types, the exposure of the environment to molybdenum is regarded as significant.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
water	LOW	LOW

Bioaccumulative potential

Ingredient	Bioaccumulation
water	LOW (LogKOW = -1.38)
ammonium heptamolybdate	LOW (BCF = 5.7)

Mobility in soil

Ingredient	Mobility
water	LOW (KOC = 14.3)

SECTION 13 DISPOSAL CONSIDERATIONS

Waste treatment methods

Product / Packaging disposal

Dispose of according to federal, state, and local regulations.

SECTION 14 TRANSPORT INFORMATION

Labels Required



Marine Pollutant NO

Land transport (DOT)

UN number	1760
Packing group	II
UN proper shipping name	Corrosive liquids, n.o.s. (contains sulfuric acid)
Environmental hazard	No relevant data
Transport hazard class(es)	Class 8
Special precautions for user	Special provisions B2, IB2, T11, TP2, TP27

Air transport (ICAO-IATA / DGR)

UN number	1760
Packing group	1

UN proper shipping name	Corrosive liquid, n.o.s. * (contains sulfuric acid)	
Environmental hazard	No relevant data	
Transport hazard class(es)	ICAO/IATA Class 8 ICAO / IATA Subrisk Not Applicable ERG Code 8L	
Special precautions for user	Special provisions Cargo Only Packing Instructions Cargo Only Maximum Qty / Pack Passenger and Cargo Packing Instructions Passenger and Cargo Maximum Qty / Pack Passenger and Cargo Limited Quantity Packing Instructions Passenger and Cargo Limited Maximum Qty / Pack	A3A803 855 30 L 851 1 L Y840 0.5 L

Sea transport (IMDG-Code / GGVSee)

UN number	1760
Packing group	II.
UN proper shipping name	CORROSIVE LIQUID, N.O.S. (contains sulfuric acid)
Environmental hazard	Not Applicable
Transport hazard class(es)	IMDG Class 8 IMDG Subrisk Not Applicable
Special precautions for user	EMS NumberF-A , S-BSpecial provisions274Limited Quantities1 L

Transport in bulk according to Annex II of MARPOL 73 / 78 and the IBC code

IMO MARPOL 73/78 (Annex II) - List of Noxious Liquid sulfuric acid Y	Pollution Category	Ingredient	Source
Substances Carried in Bulk	Y	sulfuric acid	II) - List of Noxious Liquid

SECTION 15 REGULATORY INFORMATION

Safety, health and environmental regulations / legislation specific for the substance or mixture

water(7732-18-5) is found on the following regulatory lists	"US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory"
sulfuric acid(7664-93-9) is found on the following regulatory lists	"US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants", "US - Idaho - Limits for Air Contaminants", "US - Hawaii Air Contaminant Limits", "US - California Permissible Exposure Limits for Chemical Contaminants", "US ACGIH Threshold Limit Values (TLV) - Carcinogens", "US National Toxicology Program (NTP) 13th Report Part A Known to be Human Carcinogens", "US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants", "US - Oregon Permissible Exposure Limits (Z-1)", "US - Michigan Exposure Limits for Air Contaminants", "US - Oregon Permissible Exposure Limits (Z-1)", "US - Michigan Exposure Limits for Air Contaminants", "International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs", "US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values", "US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs)", "US NIOSH Recommended Exposure Limits (RELs)", "International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List Passenger and Cargo Aircraft", "US - Alaska Limits for Air Contaminants", "US - Washington Permissible exposure limits of air contaminants", "US - Minnesota Permissible Exposure Limits (PELs)", "US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELs)", "US ACGIH Threshold Limit Values (TLV)", "US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants", "US - Naciones Control Act (TSCA) - Chemical Substance Inventory", "US - Wooming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants", "US - New Jersey Right to Know - Special Health Hazard Substance List (SHHSL): Carcinogens", "US OSHA Permissible Exposure Levels (PELs) - Table Z1"
ammonium heptamolybdate(12027-67-7) is found on the following regulatory lists	"US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants", "US - Idaho - Limits for Air Contaminants", "US - Hawaii Air Contaminant Limits", "US - California Permissible Exposure Limits for Chemical Contaminants", "US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants", "US - Oregon Permissible Exposure Limits (Z-1)", "US - Michigan Exposure Limits for Air Contaminants", "US NOSH Recommended Exposure Limits (RELs)", "US - Alaska Limits for Air Contaminants", "US - Washington Permissible exposure limits of air contaminants", "US - Michigan Exposure Limits (RELs)", "US - Alaska Limits for Air Contaminants", "US - Washington Permissible exposure limits of air contaminants", "US - Minnesota Permissible Exposure Limits (PELs)", "US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants", "US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory", "US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants", "US OSHA Permissible Exposure Levels (PELs) - Table Z1"

SECTION 16 OTHER INFORMATION

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

A list of reference resources used to assist the committee may be found at:

www.chemwatch.net/references

The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

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