

## Phosphate (reactive, ortho) – Vanadomolybdophosphoric Acid Method

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### Applications and Industries

Domestic and industrial wastewater, industrial process waters, boiler water, cooling water, surface and ground water, potable water, seawater

Power generation, pulp and paper, food and beverage industries

### References

APHA Standard Methods, 23<sup>rd</sup> ed., Method 4500-P C - 2005  
ASTM D 515-82, Phosphorous in Water, Test Method C

### Chemistry

In an acidic solution, ortho-phosphate reacts with ammonium molybdate and ammonium vanadate to produce a yellow colored complex in direct proportion to the phosphate concentration. Results are expressed in ppm (mg/L) phosphate as PO<sub>4</sub>. To convert results from ppm PO<sub>4</sub> to ppm P, divide by 3.06.

### Available Analysis Systems

*Visual colorimetric:* CHEMets®

*Instrumental colorimetric:* Vacu-vials®

### Storage Requirements

Products should be stored in the dark and at room temperature.

### Shelf Life

*When stored in the dark and at room temperature:*

*Visual colorimetric:*

CHEMets refills, color comparators: at least 1 year

*Instrumental colorimetric:*

Vacu-vials kit: at least 1 year

### Safety Information

Safety Data Sheets (SDS) are available upon request and at [www.chemetrics.com](http://www.chemetrics.com). Read SDS before using these products. Breaking the tip of an ampoule in air rather than water may cause the glass ampoule to shatter. Wear safety glasses and protective gloves.

### Interference Information

- High concentrations of ferrous iron and other reducing agents may cause blue color development rather than yellow.
- Molybdate at >1000 ppm, thiosulfate, sulfide, thiocyanate, bismuth, thorium, fluoride, or arsenate may cause a false negative result.
- Arsenate and silica can become positive interferences only if the sample is heated.
- Samples with extreme pHs or that are highly buffered should be adjusted to approximately 7 prior to analysis.
- Unexpectedly high results may reflect sample contamination from labware. If contamination is suspected, labware can be rinsed with dilute sulfuric acid followed by distilled water.
- Condensed phosphates (pyro-, meta-, and other polyphosphates) and organically bound phosphates do not respond to this test.
- Sample color or turbidity may make a color match difficult during visual colorimetric testing and may cause a false positive result with instrumental colorimetric tests. CHEMetrics' Sample Zeroing Accessory Pack can be used to correct for potential errors during instrumental analysis.

### Accuracy Statement

*Statements of accuracy are based on laboratory tests performed under ideal testing conditions using standards of known concentration prepared in deionized water.*

*CHEMets kits:*

± 1 color standard increment

*Vacu-vials kit:*

<1.3 ppm at 0 ppm  
±1.5 ppm at 5.0 ppm  
±4.0 ppm at 20.0 ppm  
±6.0 ppm at 60.0 ppm